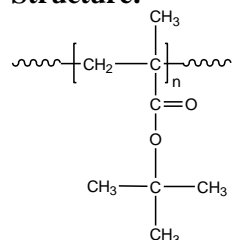


Sample Name: Poly(t-butyl methacrylate)  
*Atactic rich*

Sample #: P11323A-tBuMA

**Structure:**



**Composition:**

$M_n \times 10^3$	PDI
120.0	1.4
S:H:I	0:92:8

**Synthesis Procedure:**

Poly(t-butyl methacrylate) is obtained by RAFT polymerization process.

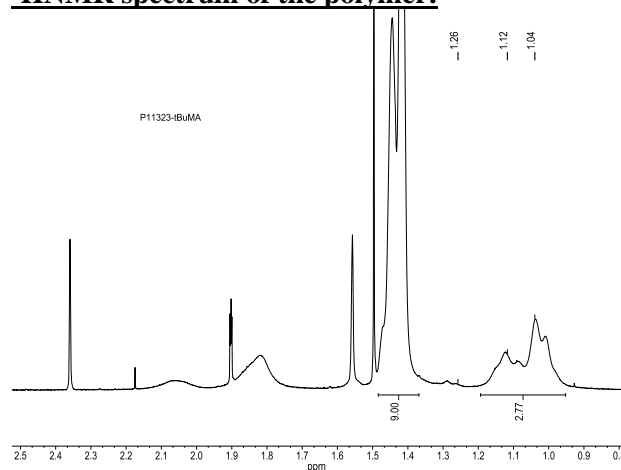
**Characterization:**

The product was characterized by size exclusion chromatography (SEC) and  $^1\text{H}$  NMR.

**Solubility:**

Poly(tert butylmethacrylate) is soluble in THF,  $\text{CHCl}_3$ , toluene and dioxane. The polymer precipitates from cold methanol and ethanol.

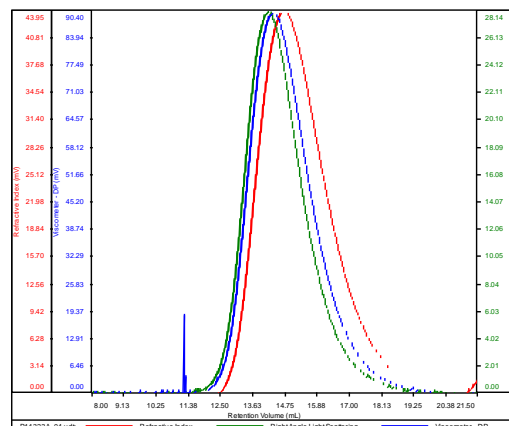
**$^1\text{H}$ NMR spectrum of the polymer:**



**SEC elugram of Homopolymer:**

P11323A-tBuMA

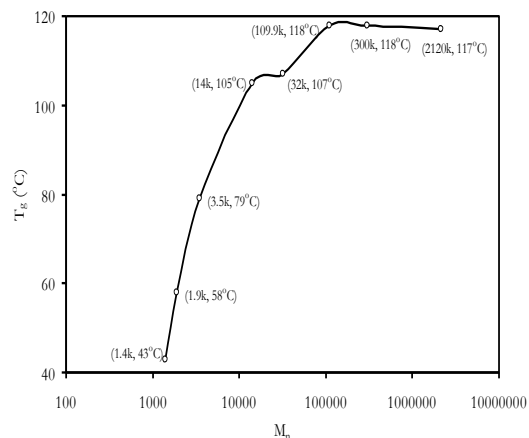
Conc	13.9503
dn/dc	0.0650
Solvent	DMF w 0.023M LiBr
Flow Rate	0.7000
Method	PS80k-May2017-0000.vcm



Sample	Mn	Mw	Mp	Mw/Mn	IV
P11323A_01.vdt	119,936	170,910	162,277	1.425	0.1814

**DSC thermogram of the Product:**

Tg of poly t-butyl methacrylate as function of molecular weight



**Tg vs MW for selected poly t-butyl methacrylate**

$M_n \times 10^3$	$T_g$ (°C)	$M_n \times 10^3$	$T_g$ (°C)
1.4	43	32	107
1.9	58	109.9	118
3.5	79	300	118
14	105	2120	117

**References for further information:**

S. K. Varshney, Z. Gao, Xing Fu Zhong, A. Eisenberg

“Effect of Lithium Chloride on the “Living” Polymerization of tert-Butylmethacrylate and Polymer Microstructure Using Monofunctional Initiators” Macromolecules, 1994, 27, 1076.