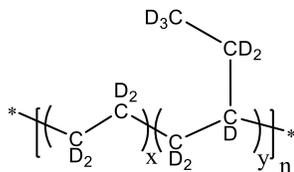


Sample Name:
Deuterated Poly(ethylene-d₄-co-butylene-d₈)

Sample #: **P40895-dEB**

Structure:



Composition:

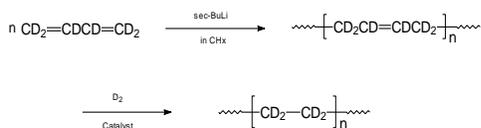
$M_n \times 10^3$ (g/mol)	M_w/M_n
175.0	1.2

Thermal properties:

Melting point, T_m	Crystallization point, T_{cr}
49 °C	37-44 °C

Synthesis procedure:

Deuterated poly(ethylene-co-butylene) was obtained by deuteration of poly(1,4-butadiene-d₆), which was synthesized by living anionic polymerization of butadiene-d₆ in non-polar solvent. The scheme of reaction is presented below:



Characterization:

Deuterium NMR spectroscopy was used to confirm the structure of polybutadiene-d₆ rich in 1,4-addition.

The complete deuteration of the product was confirmed by FT-IR spectroscopy analysis by disappearance of alkene double bond (C=C at 971 cm⁻¹).

The molecular weight and polydispersity index were obtained by size exclusion chromatography (SEC) of poly(1,4-butadiene-d₆) precursor using THF as an eluent; and the molecular weight of polyethylene-d₄ was calculated accordingly.

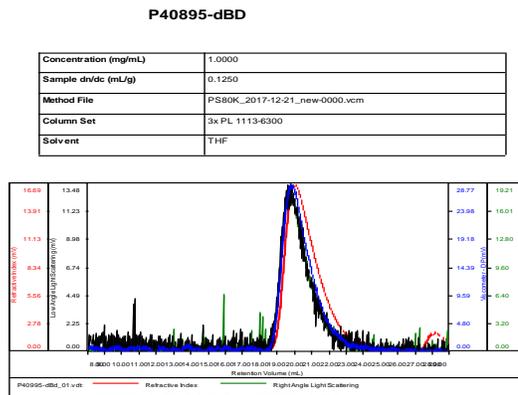
Thermal analysis was performed on TA Instruments Q100 differential scanning calorimeter (DSC) under a nitrogen atmosphere at a scan rate 10 °C/min.

Solubility:

The product is soluble in hot toluene and xylene. The obtained solution has light ivory color; this coloration is due to the presence of trace amount (we expect <5–6

ppm) of the Wilkinson catalyst used in synthesis (and which is hard to remove from the final product).

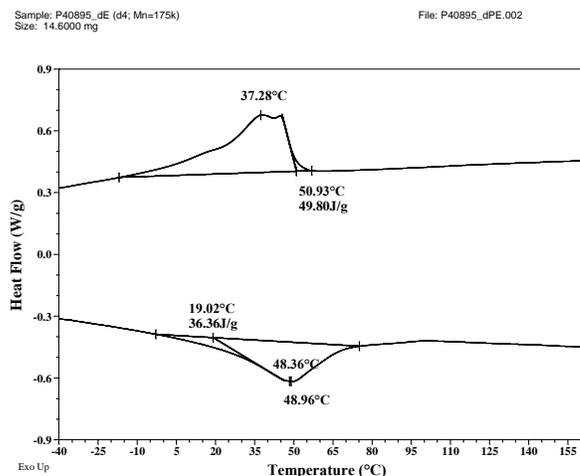
SEC chromatogram of dPBd precursor:



Sample	Mn (Da)	Mw (Da)	Mw/Mn	IV (dL/g)	Mp (Da)
P40995-dBd_01.v dt	165,783	198,982	1.200	1.5054	213,586

DSC thermograms of the dPE product:

1st cooling (upper) and 2nd heating (lower) scans, both performed at a rate 10 °C/min.:



FT-IR spectra of dPE (top) and dPBd (bottom):

