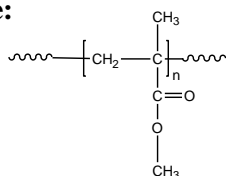


**Sample Name: Poly (methyl methacrylate)**

*Different microstructure*

**Sample #: P43238-MMA**

**Structure:**



**Composition:**

$M_n \times 10^3$	PDI
450.0	1.17

Isotactic: Hetero : Syndio	4:28:68
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**Synthesis Procedure:**

Poly (methyl methacrylate) is obtained by anionic polymerization using Cumyl Potassium initiator.

**Characterization:**

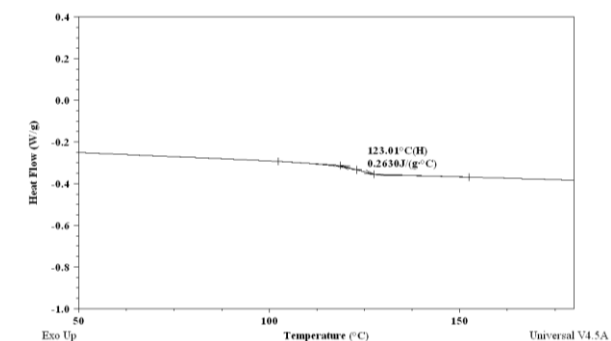
The product was characterized by size exclusion chromatography (SEC) and  $^1\text{H}$  NMR.

**Solubility:**

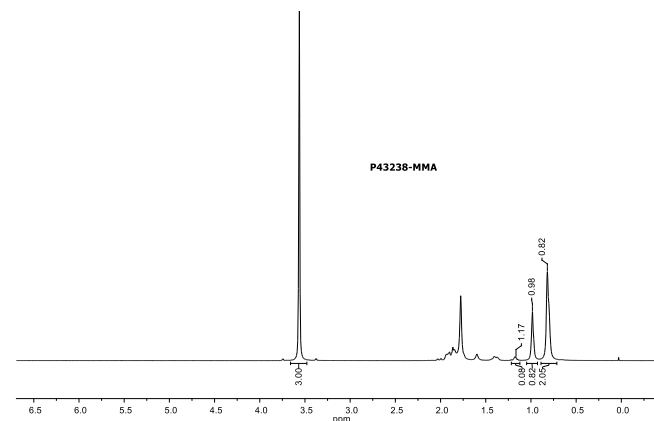
The polymer is soluble in THF,  $\text{CHCl}_3$ , toluene and dioxane. The polymer precipitates from cold methanol and ethanol.

**$T_g$  vs MW for selected atactic PMMA:**

$M_n \times 10^3$	$T_g$ ( $^{\circ}\text{C}$ )	$M_n \times 10^3$	$T_g$ ( $^{\circ}\text{C}$ )
1.1	51	36	98
2.5	76	55	111
5.0	91	70	107
15	101	127	115
19	107	230	114
29	96	700	121



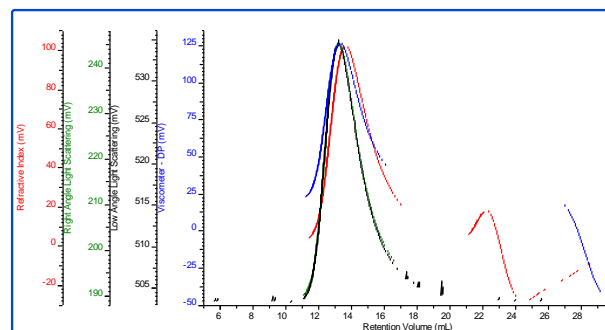
**$^1\text{H}$  NMR spectrum of PMMA Sample:**



**SEC elugram of PMMA homopolymer:**

**P43238-MMA**

dn/dc	0.0570
Flow Rate	0.7000
Solvent	DMF with LiBr
Method	Calibration_2020-11-25_PMMA-85K-0003.vcm



Sample	$M_n$	$M_w$	$M_p$	$M_w/M_n$
P43238_1_2021-06-02	449,729	529,657	515,019	1.178

**DSC thermogram:**

T<sub>g</sub> of atactic poly methyl methacrylate as function of molecular weight

