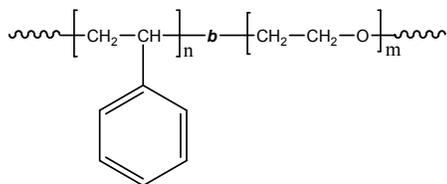


Sample Name: **Poly(styrene-b-ethylene oxide)**

Sample #: **P11155B-SEO**

Structure:



Composition:

$M_n \times 10^3$ S-b-EO	PDI
20.5-b-11.5	1.05

Synthesis Procedure:

Poly(styrene-b-ethylene oxide) diblock copolymer is prepared by living anionic polymerization.

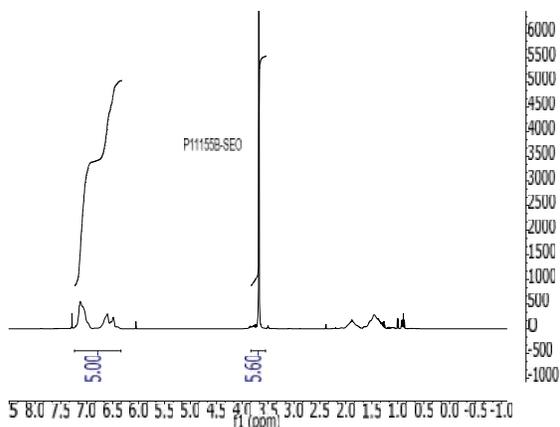
Characterization:

The molecular weight and polydispersity index (PDI) of the block copolymer are characterized by size exclusion chromatography (SEC). The composition of the block copolymer was calculated from  $^1H$ -NMR by comparing the peak area of the phenyl polystyrene protons between 6.4 to 7.2 ppm and the ethylene oxide protons at 3.65 ppm.

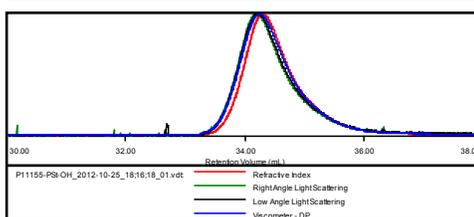
Solubility:

The polymer is soluble in THF (at 35 °C),  $CHCl_3$ , benzene, toluene, dioxane. Low molecular weight SEO with high contents of the polyethylene oxide block can also be solubilized in methanol and water.

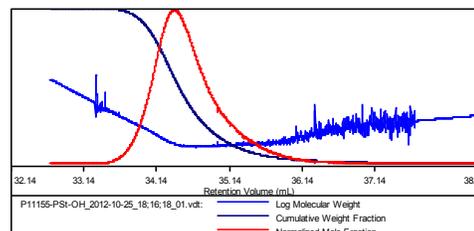
$^1H$  NMR spectrum of the sample



Concentration (mg/mL)	9.0852
Sample dn/dc (mL/g)	0.1850
Method File	PS80K-Oct-2012-0002.vcm
Column Set	3x PL 1113-6300
System	System 1

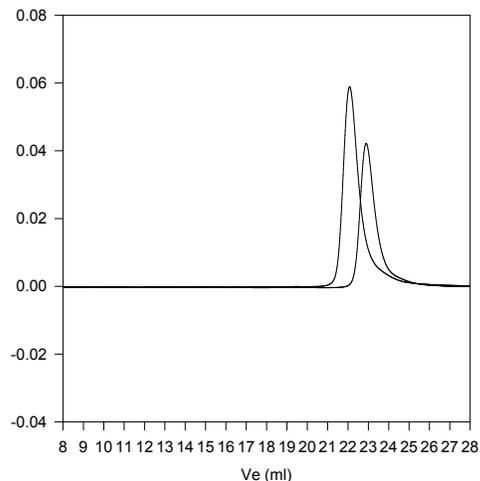


Sample	Mn (Da)	Mw (Da)	Mp (Da)	Mw/Mn	IV (dL/g)
P11155-PS-OH_2012-10-25_18;16;18_01.vd	20,538	21,761	19,890	1.060	0.2138



**SEC profile of the block copolymer:**

**P11155B-SEO**



— Poly(styrene),  $M_n=20,500$ ,  $M_w=21,100$ ,  $PI=1.06$

— Block Copolymer  $PS(20,500)$ -b- $PEO(11,500)$ ,  $PI=1.05$

## Thermal analysis of the sample# P11155B-SEO

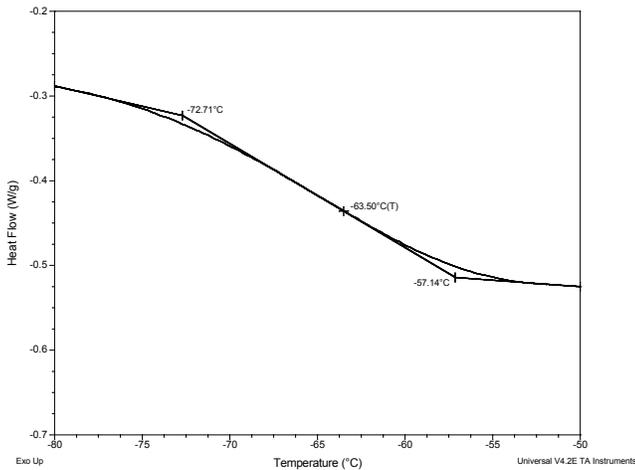
Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 20°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature ( $T_g$ ).

## Thermal analysis results at a glance

<b>For PS block <math>T_g</math>: 85°C</b>		
<b>For PEO block</b>		
$T_g$ : -63°C	$T_m$ : 37°C	$T_c$ : -18 & -45°C

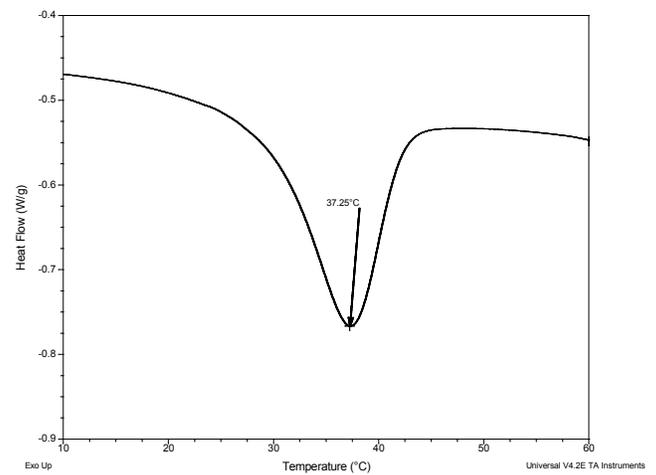
## Thermogram for the sample

### For PEO block:



## Melting and crystallization curve for the PEO block

The melting temperature ( $T_m$ ) was taken as the maximum of the endothermic peak where as the crystallization temperature ( $T_c$ ) was considered as the minimum of the exothermic peak.



### For PS block

