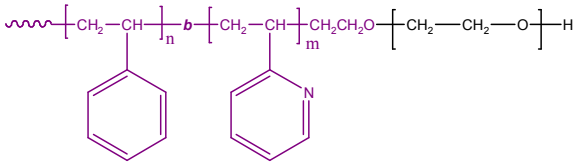
Sample Name:

Poly(styrene-b-2-vinyl pyridine-ethylene oxide)

Sample #: P18206P-S2VPEO

Structure:



Composition:

Mn x 10^3 S-b-2VP-b-EO	PDI
32.3-b-29.3-b-11.5 Calculated from ^{H}NMR	1.11

Synthesis Procedure:

Poly(styrene-b-2-vinyl pyridine-ethylene oxide) triblock copolymer is prepared by living anionic polymerization by successive addition of monomer using cumyl potassium as initiator.

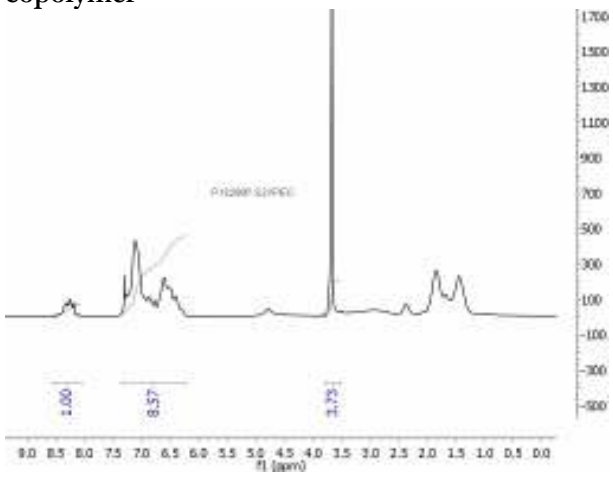
Characterization:

Polymer at different stages of polymerization was analyzed by size exclusion chromatography (SEC).The Block copolymer composition was then calculated from ^{1}H-NMR spectroscopy .

Solubility:

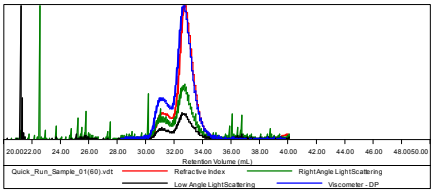
Poly(styrene-b-2 vinylpyridine-b-ethylene oxide) is soluble in THF, toluene, and $CHCl_3$.

^{1}H-NMR Spectrum of the polymer S2 VPEO triblock copolymer

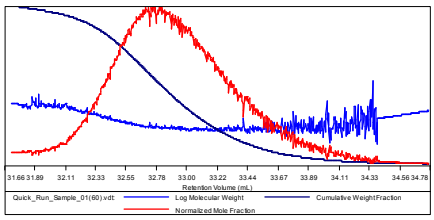


Sample ID: P18206-S

Concentration (mg/mL)	2.8547
Sample dn/dc (mL/g)	0.1850
Method File	PS80K-Sep26-2013-0000.vcm
Column Set	3x PL 1113-6300
System	System 1

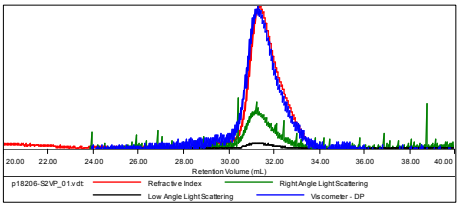


Sample	Mn	Mw	Mp	Mw/Mn	IV
Quick_Run_Sample_01(60).vdt	32,279	33,834	30,311	1.048	0.2549

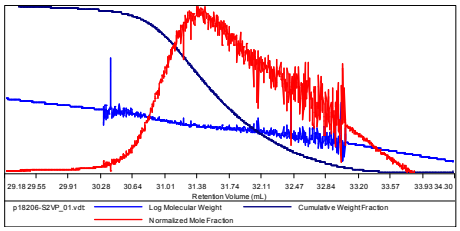


Sample ID: PP18206-S2VP

Concentration (mg/mL)	0.9318
Sample dn/dc (mL/g)	0.1650
Method File	PS80K-Sep26-2013-0000.vcm
Column Set	3x PL 1113-6300
System	System 1



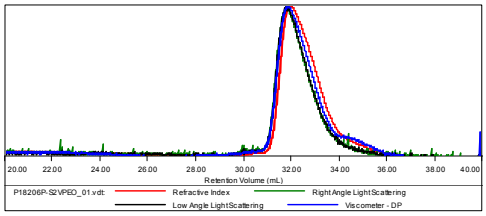
Sample	Mn	Mw	Mp	Mw/Mn	IV
p18206-S2VP_01.vdt	61,403	68,956	67,899	1.123	0.3614



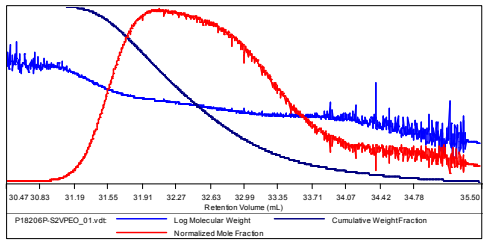
SEC for the triblock polymer:

Sample ID: P18206P-S2VPEO

Concentration (mg/mL)	6.4237
Sample dn/dc (mL/g)	0.1250
Method File	PS80K-Sep26-2013-0000.vcm
Column Set	3x PL 1113-6300
System	System 1



Sample	Mn	Mw	Mp	Mw/Mn	IV
P18206P-S2VPEO_01.vdt	74,599	82,044	90,053	1.100	0.3217



References:

1. S. K. Varshney, X. F. Zhong and A. Eisenberg *Macromolecules* **1993**, 26, 701-706.

2. Gohy J.-F., Willet N., , Zhang J.-X., Varshney S., Jerome, ., *pH dependence of the morphology of aqueous micelles formed by poly(styrene)-block-poly(2-vinylpyridine)-block-poly(ethylene oxide) copolymers,* e-polymers 2002, 35.

3. Gohy, J.-F., Lohmeijer, B Varshney S,K, Decamps B., Leroy E., Boileau S., Schubert U. S., *Stimuli-responsive aqueous micelles from an ABC metallo-supramolecular triblock copolymer*, Macromolecules 2002, 35, 9748-9755.

4. Gohy, J.-F., Mores S., Varshney S. K., Jerome, R., *Self-organization of water-soluble complexes of a poly(2-vinylpyridinium)-block-poly(ethylene oxide) diblock and a fluorinated anionic surfactant*, Macromolecules 2003, 36, 2579-2581.

5. Leil L., Gohy J.-F., Willet N., , Zhang J.-X., Varshney S., Jerome R., *Tuning of the morphology of core-shell-corona aqueous micelles: I. sphere–to-cylinder transition*, Macromolecules 2004, 37, 1089-1094.

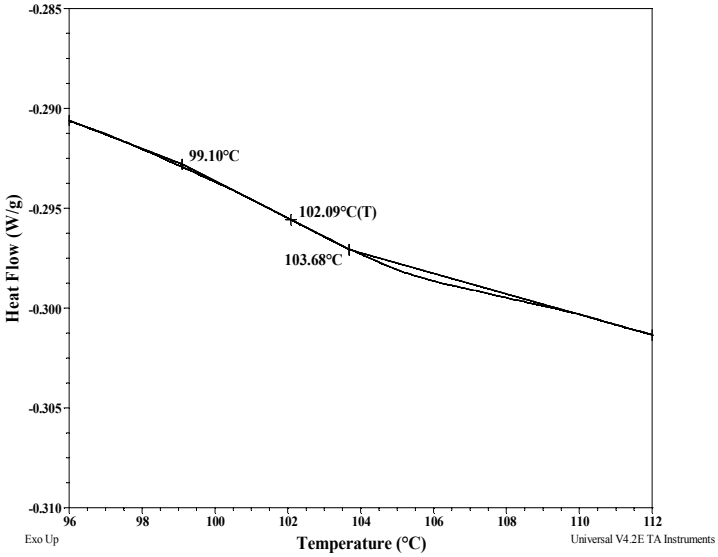
Thermal Analysis of the sample S2VPEO

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min.The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature (T_g).

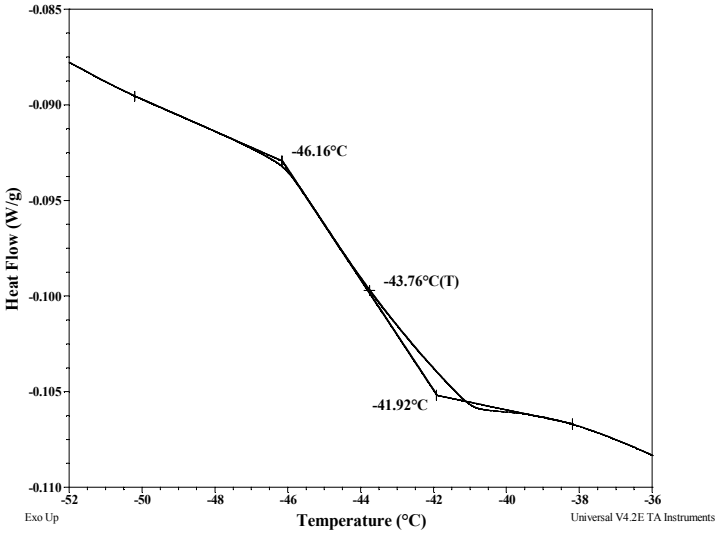
Thermal analysis results at a glance

For PS block: T_g: 102°C		**For 2VP block**: T_g: Not distinct
For PEO block		
T_g: -44°C	T_m: 61°C	T_c: 34°C

Thermogram for PS block:



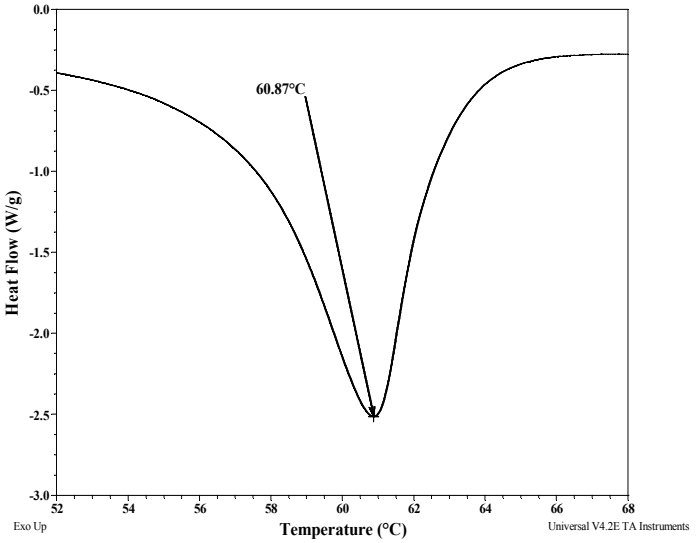
Thermogram for PEO block:



Melting and crystallization curve for the sample

The melting temperature (T_m) was taken as the maximum of the endothermic peak where as the crystallization temperature (T_c) was considered as the minimum of the exothermic peak.

Melting curve for PEO block



Crystallization curve For PEO block