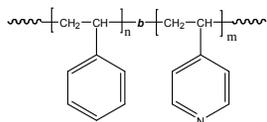


Sample Name: Poly(styrene-b-4-vinyl pyridine)

Electronic grade

Sample #: P5560P-S4VP

Structure:



Composition:

Mn x 10 ³ PS-b-4VP	PDI
311.0-b-120.0	1.2
T _g for PS block: 107°C	T _g for 4VP block: 154°C

Synthesis Procedure:

Poly(styrene-b-4-vinyl pyridine) is prepared by living anionic polymerization in THF or THF-DMF solvent mixtures at -78 °C. Polystyrene macroanions were end capped with a unit of diphenyl ethylene (DPE) before adding 4-vinylpyridine (4VP) monomer. For further details please see our published articles.^{1,2}

Characterization:

An aliquot of the anionic polystyrene block was terminated before addition of 4-vinyl pyridine and analyzed by size exclusion chromatography (SEC) in DMF to obtain the molecular weight and polydispersity index (PDI). The block copolymer composition was then calculated from ¹H-NMR spectroscopy by comparing the peak area of the two aromatic 4-VP protons at about 8.5 ppm with the peak area of the aromatic protons of polystyrene at 6.3-7.2 ppm. The composition of the block copolymer can also be determined by titration in acetic acid/HClO₄ using crystal violet indicator. Copolymer PDI is determined by SEC.

Difficulties in determination of chemical compositions of such high molecular weights product:

These high molecular weights polymer by HNMR in CdCl₃ do not gives the correct compositions. These polymers were characterized by HNMR in DMF at room temperature and at 50°C. In CdCl₃, HNMR chemical shift occurs at 8.4 ppm. The product compositions were verified by titration in acetic acid using crystal violet and HClO₄ (per chloric acid) acid-base titration. The Compositions were also verified by FTIR taking the comparison of styrene characteristics at 3059 cm⁻¹ and for 4VP at 1556 Cm⁻¹. From titration it shows the **presence of 1.5 wt % 4VP component.**

Thermal analysis of the samples was carried out using a differential scanning calorimeter (TA Q100) at a heating rate of 10°C/min. The inflection glass transition temperature (T_g) of the sample has been considered.

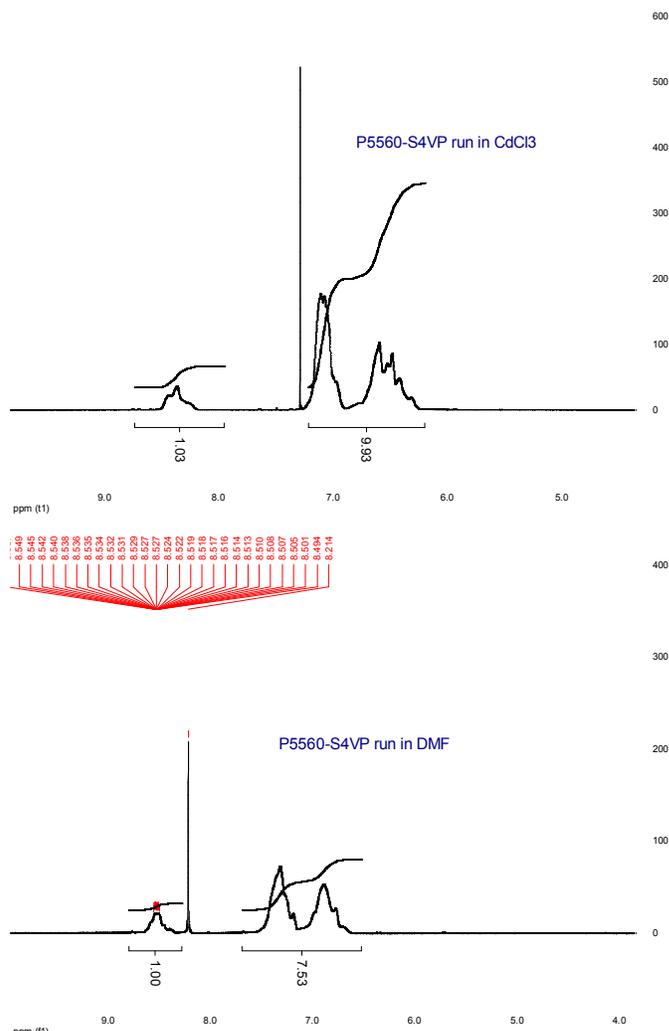
Solubility:

Poly(styrene-b-4-vinyl pyridine) is soluble in DMF, CHCl₃.

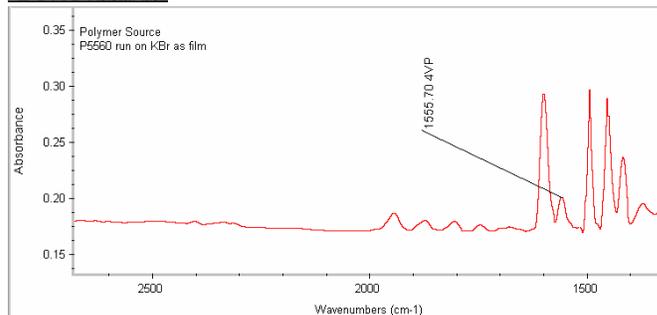
Purification of the obtained polymer was carried out rigorously as follows to ensure the removal of the catalyst side product:

1. First the polymer soxhlet in Cyclohexane to remove unreacted homopolystyrene fraction.
2. Dissolved the polymer in CHCl₃ and wash with de-ionized distilled water to remove the any soluble organic catalyst side product.
3. Polymer extracted from water with chloroform.
4. Polymer solution in CHCl₃ was dried over anhydrous sodium sulfate.
5. Solution filtered and than passed through a column packed with basic Al₂O₃.
6. Solution concentrated on rota-evaporator
7. Solution precipitated in cold hexane and redissolved in benzene and freeze dried.
8. Final dried under vacuum for 48h at 50°C.

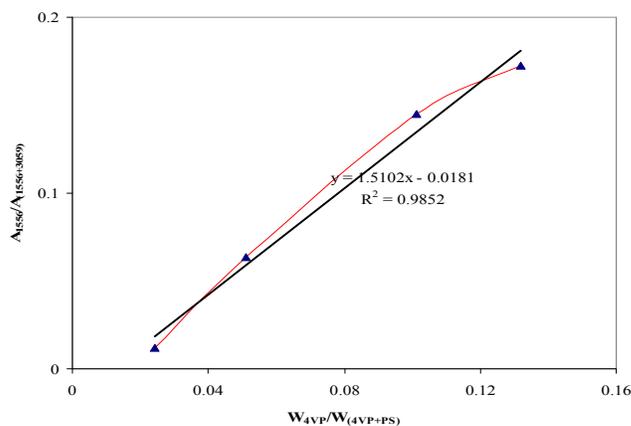
¹H-NMR Spectrum of Sample



FTIR Results:

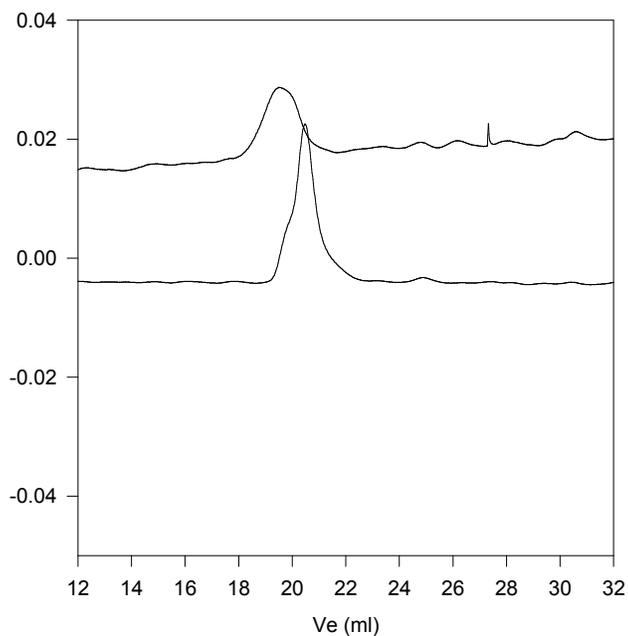


Relationship between weight fraction and area fraction of PS and 4VP



SEC of Sample:

P5560-S4VP

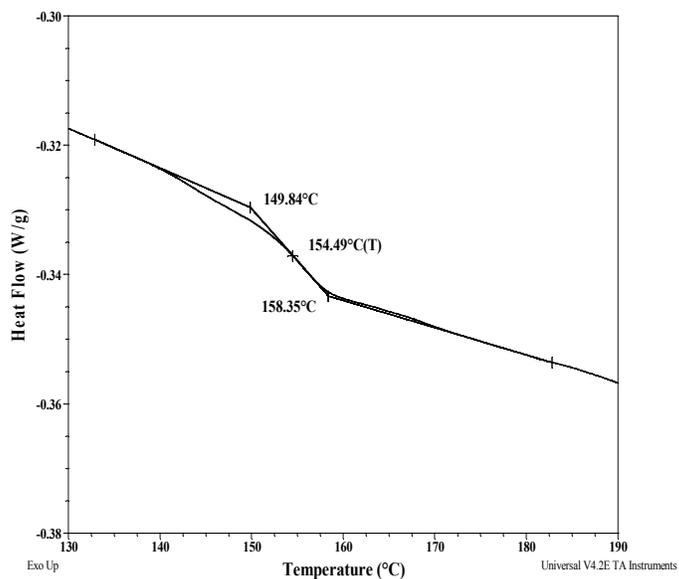


Size exclusion chromatography of P(s-b-4VP) in DMF at 40 °C:

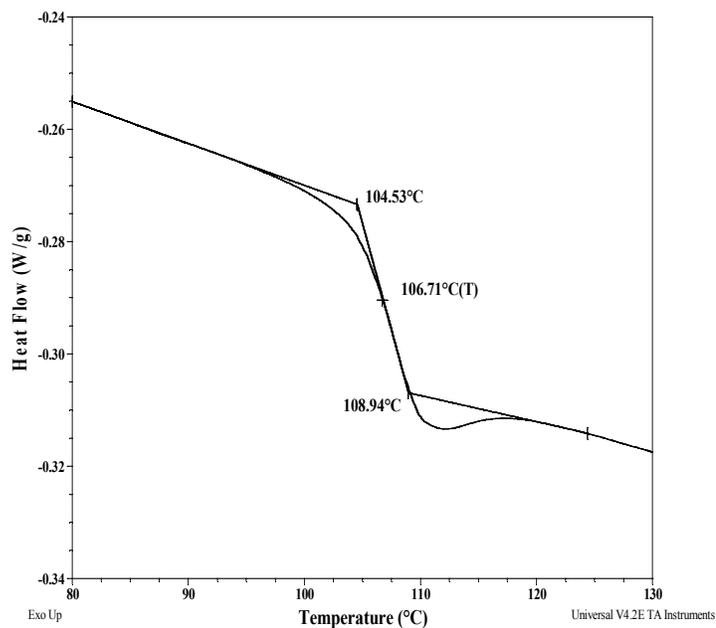
— PS block: $M_n=311,000$, $M_w=367,000$, $PI=1.15$

- - - Block Copolymer PS-4VP (311,000)-b-4VP(120,000), $PI=1.20$

DSC thermogram for 4VP block:



DSC thermogram for PS block:



References:

- (1). S. K. Varshney, X. F. Zhong and A. Eisenberg, *Macromolecules*, **1993**, 26, 701-706.
- (2). Z.Gao, S. K. Varshney, S. Wong, A. Eisenberg *Macromolecules*, **1994**, 27, 7923-7927.