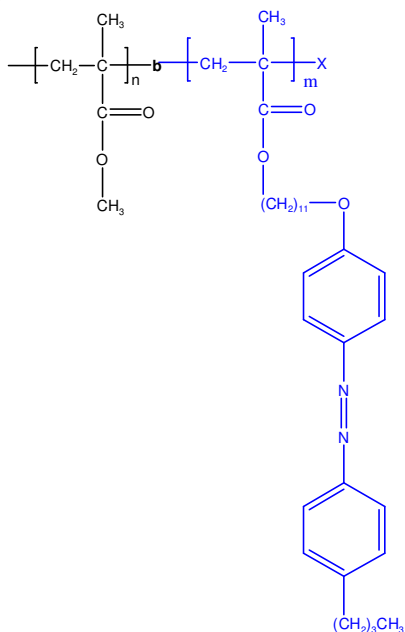


**Sample Name: Poly(Methylmethacrylate-b-AzoMA)**  
*AzoMA=11-[4-(4-butylphenylazo)phenoxy]-undecyl methacrylate)*

**Sample #: P5853-MMAAzoMA**

**Structure:**

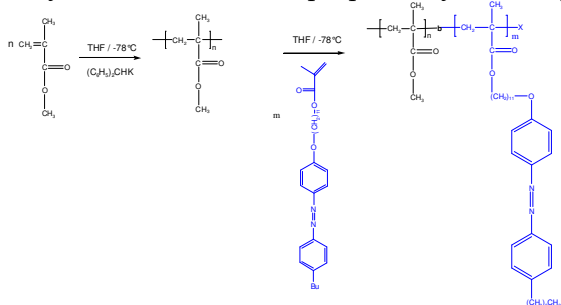


**Composition:**

Mn x 10 <sup>3</sup> PMMA-b-PAzoMA	PDI
9.0-b-2.0	1.2

**Synthesis Procedure:**

Poly(MMA-b-AzoMA) is prepared by anionic process:



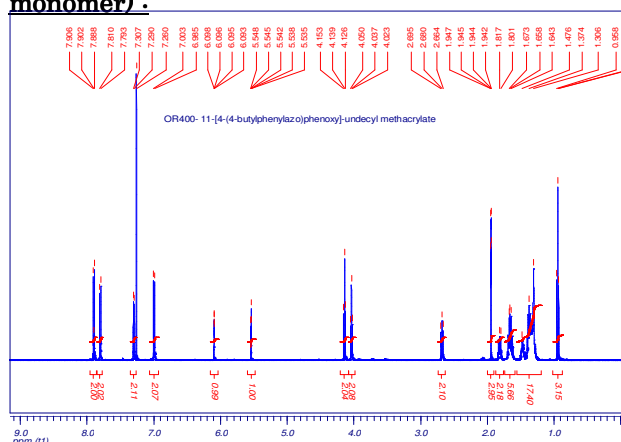
**Characterization:**

Block copolymer were analyzed by size exclusion chromatography (SEC) to obtain the molecular weight. The final block copolymer composition was calculated from <sup>1</sup>H-NMR spectroscopy.

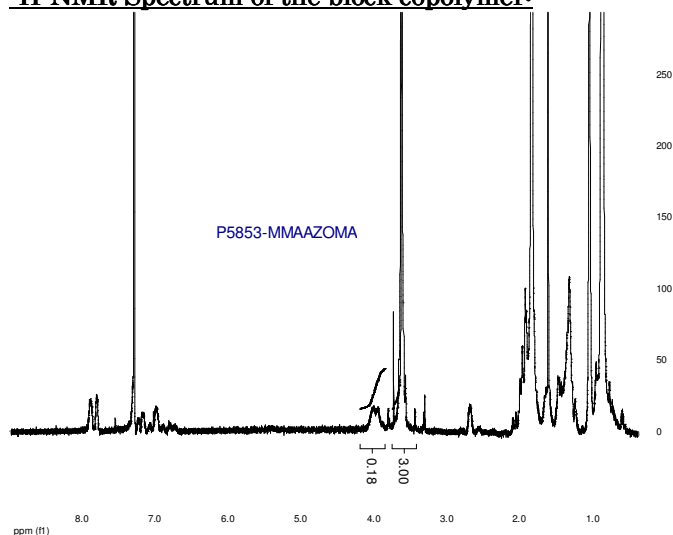
**Solubility:**

Poly(MMA-b-AzoMA) is soluble in THF, acetone, and chloroform and it precipitates out in hexane or cold methanol.

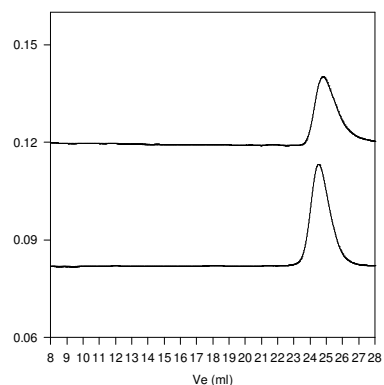
**<sup>1</sup>H-NMR Spectrum of the Azo-MA (Liquid crystalline monomer):**



**<sup>1</sup>H-NMR Spectrum of the block copolymer:**



**P5853-MMAAZOMA**



Size exclusion chromatography of poly(methyl methacrylate-b-azo MA)  
 — Poly methyl methacrylate, M<sub>n</sub>=9000, M<sub>w</sub>=10600, PI=1.18  
 — Block Copolymer PMMA(9000)-b-AzoMA(2000), PI=1.2

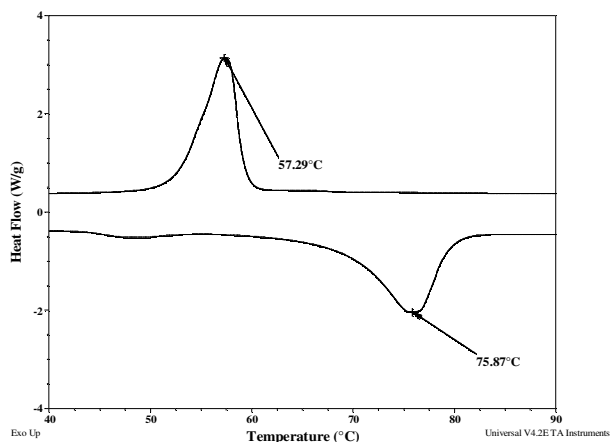
### Thermal analysis for sample#P5853-MMAzoMA

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature ( $T_g$ ).

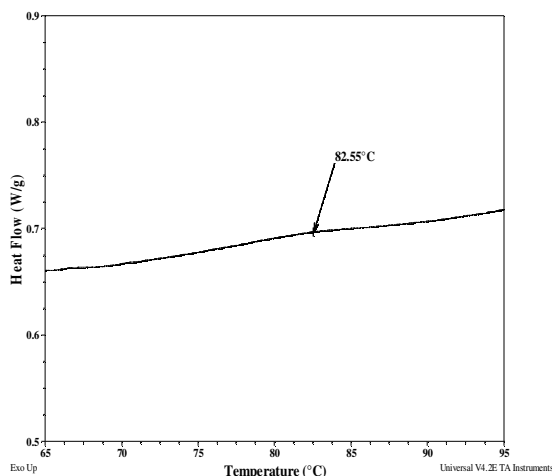
#### Melting and crystallization curve for the sample

The melting temperature ( $T_m$ ) was taken as the maximum of the endothermic peak whereas the crystallization temperature ( $T_c$ ) was considered as the minimum of the exothermic peak.

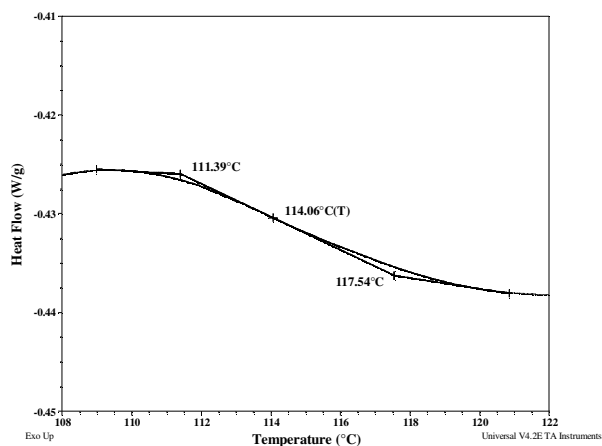
#### Thermograms for AzoMA monomer:



#### Crystallization peak for AzoMA block:



#### Thermogram for MMA block:



#### Thermal analysis results at a glance

Sample	$T_m$ (°C)	$T_c$ (°C)	$T_g$ (°C)
AzoMA monomer	76	57	
PAzoMA:	83	94	-
MMA block:	-	-	114

#### Melting curve for AzoMA block:

