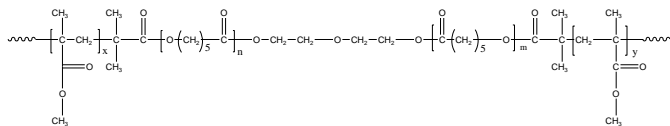


Sample Name:

Poly(methyl methacrylate -b- ε-caprolactone -b- methyl methacrylate)

Sample #: P7125- MMACLMMA**Structure:****Composition:**

Mn x 10 ³	PDI
PMMA-b-PCL-b-PMMA	
5.6-0.9-5.6	1.5

Synthesis Procedure:

Poly(methyl methacrylate -b- ε-caprolactone) -b- methyl methacrylate) is prepared by ring opening polymerization of ε-caprolactone and coordination ATRP polymerization of methyl methacrylate.

Characterization:

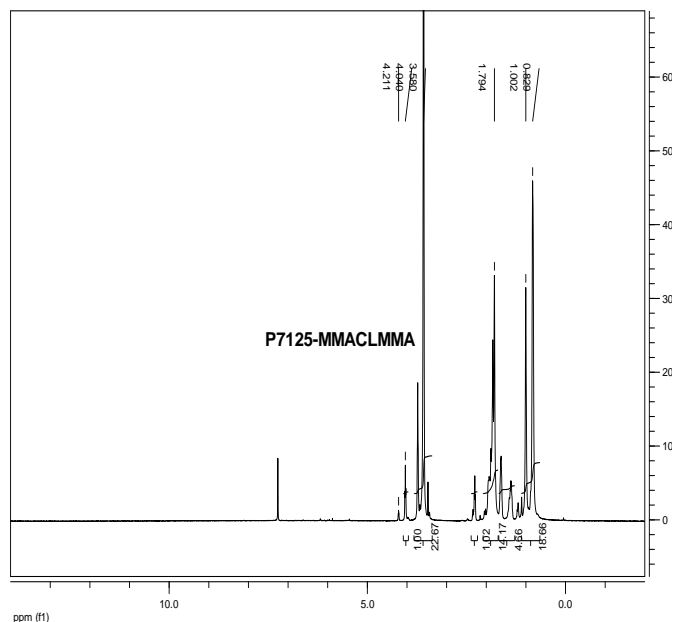
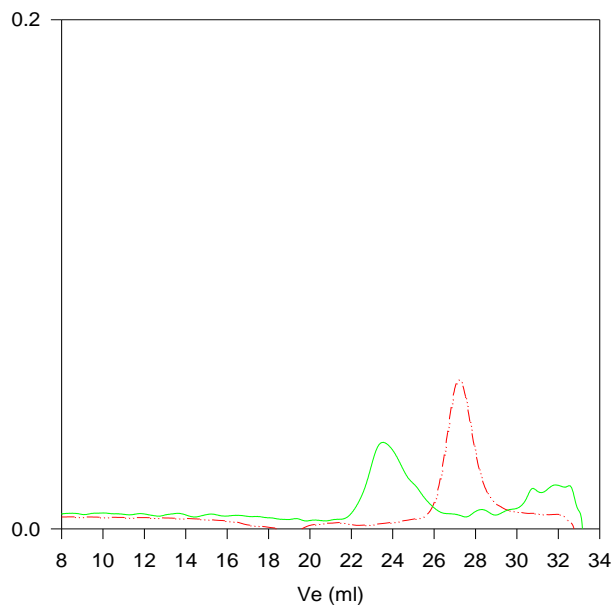
The Mn of poly(ε-caprolactone) and poly(methyl methacrylate -b- ε-caprolactone -b- methyl methacrylate) is calculated from ¹H-NMR spectroscopy by comparing the peak area of the ethylene oxide protons at about about 3.6 ppm, the ε-caprolactone protons at about 4.1 ppm and the methyl methacrylate protons at 1.9 ppm. The polydispersity index (PDI) is analyzed by size exclusion chromatography (SEC).

Thermal analysis:

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature (T_g).

Solubility:

poly(methyl methacrylate -b- ε-caprolactone -b- methyl methacrylate) is soluble in CHCl₃, THF, DMF, toluene and precipitated out from cold ethanol, diethyl ether.

¹H-NMR Spectrum of the block copolymer:**SEC of the block copolymer:****P7125- MMACLMMA**

Size exclusion chromatography:

--- Dihydroxyl ended poly(caprolactone), M_n=900, M_w=1100, PI=1.2

— Block Copolymer PMMA(5600)-PCL(900)-b-PMMA(5600), PI=1.5

Composition from ¹H NMR

Dp: MMA (56 units)-CL(8 units)-b-MMA (56 units)

Thermal analysis of the P7125 MMA CLMMA sample

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 20°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature (T_g).

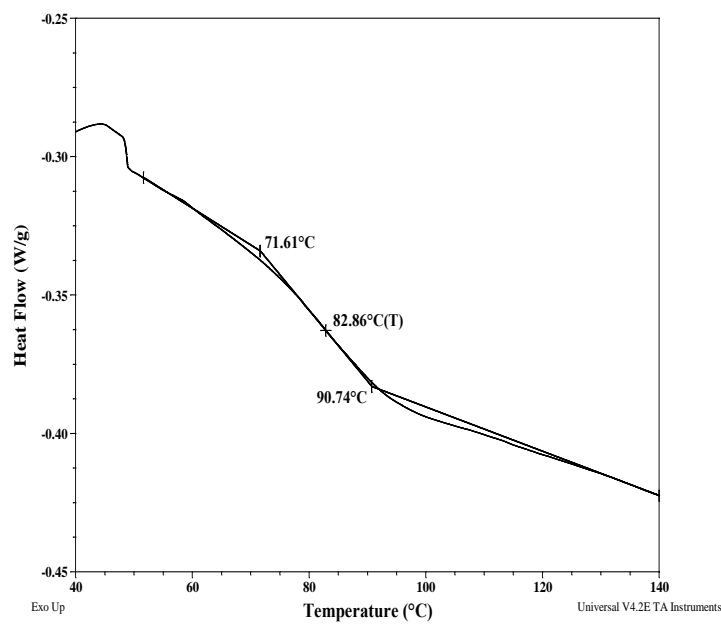
Melting and crystallization curve for the sample

The melting temperature (T_m) was taken as the maximum of the endothermic peak where as the crystallization temperature (T_c) was considered as the minimum of the exothermic peak.

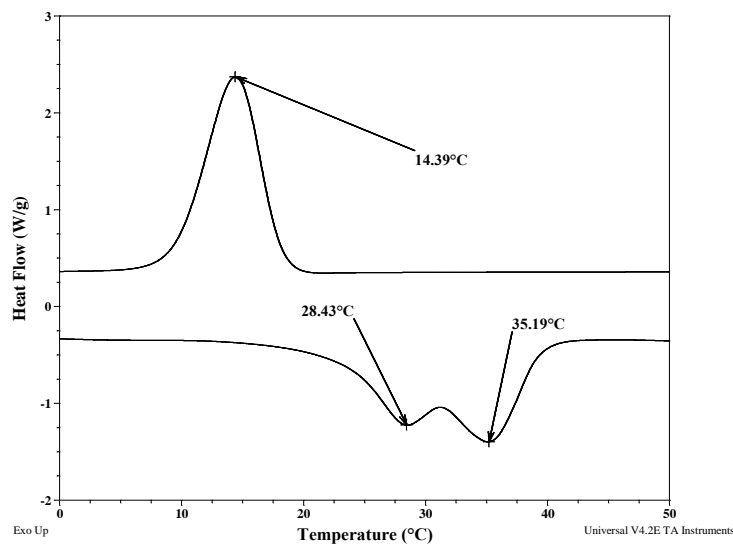
Typical thermal analysis results at a glance

Sample	T_m (°C)	T_c (°C)	T_g (°C)
MMA ($M_n=9000$)	-	-	107
MMA in triblock	-	-	83
ϵ -CL ($M_n=900$)	28 & 35	15	-64
CL in triblock	27 & 34	13	Not distinct

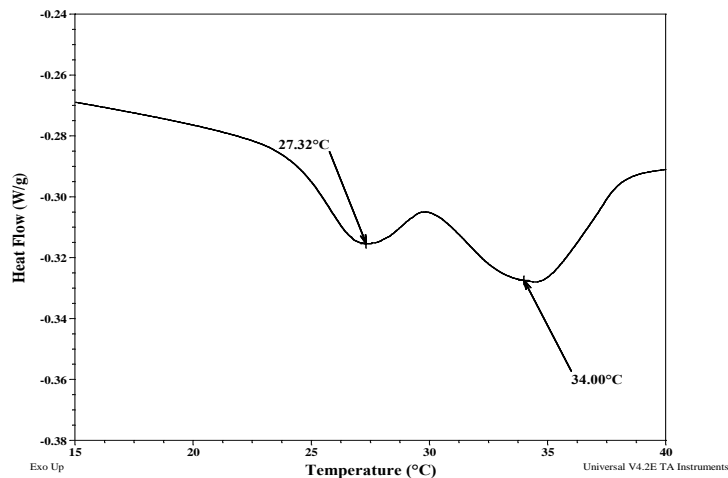
Thermogram for the MMA block in the triblock:



Thermogram of ϵ -caprolactone ($M_n \approx 900$)



Melting curve for CL block in triblock polymer:



Crystallization curve for CL block in the triblock:

