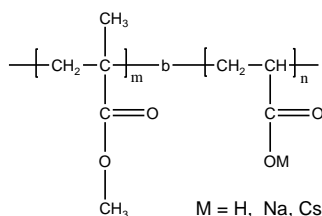


**Sample Name:** Poly(methyl methacrylate-b-acrylic acid)

**Sample #:** P8253A-MMAAA

**Structure:**



**Composition:**

$M_n \times 10^3$ PMMA-b-PAA	PDI
12.0-b-2.0	1.10
$T_g$ (°C) for MMA block: 125	$T_g$ (°C) for AA block: Not distinct

**Synthesis Procedure:**

Poly(methyl methacrylate-b-acrylic acid) is prepared by living anionic polymerization with sequence addition of methyl methacrylate followed by t-butyl acrylate or vice versa and hydrolysis of the t-butyl group.

**Characterization:**

An aliquot of the anionic poly(methyl methacrylate) block was terminated before addition of t-butyl acrylate and analyzed by size exclusion chromatography (SEC) to obtain the molecular weight and polydispersity index (PDI). The final block copolymer composition was calculated from  $^1H$ -NMR spectroscopy by comparing the peak area of the t-butyl methacrylate protons at 1.43 ppm with the peak area of the methyl methacrylate protons at 3.6 ppm. Copolymer PDI is determined by SEC.

**Thermal analysis:**

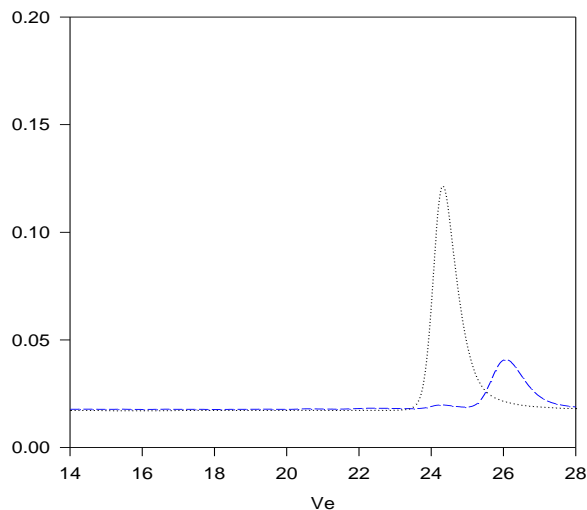
Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature ( $T_g$ ).

**Solubility:**

Poly(methyl methacrylate -b- acrylic acid) is soluble in methanol depending on the compositions. It is precipitated out from ether and hexane.

**SEC of the block copolymer:**

**P8253-MMAAtBuA Precursor for P8253A-MMAAA**

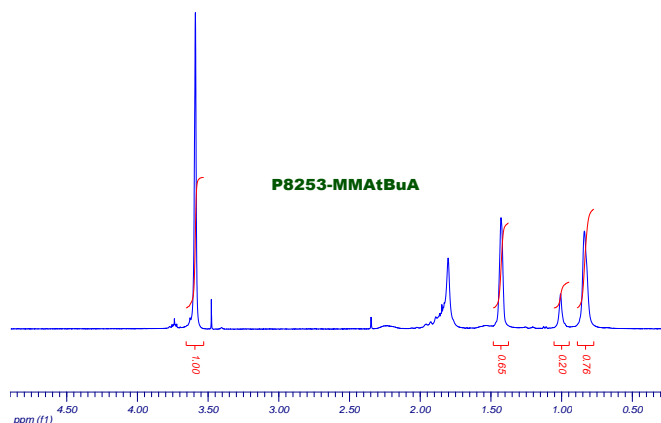


Size Exclusion Chromatography :

--- Poly tert.butylacrylate,  $M_n=3500$   $M_w/M_n=1.12$

..... Block Copolymer PMMA(12000)-tBuA(3500),  $M_w/M_n=1.10$   
After Hydrolysis of tert. butyl ester: PMMA-b-AA:12000-b-2000

**NMR of the block copolymer:**



**DSC thermogram for MMA block:**

