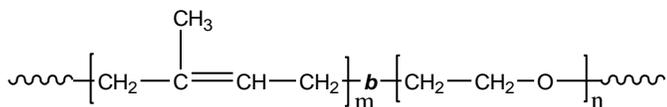


Sample Name: Poly(Isoprene-b-ethylene oxide)

Sample #: P8339-IPEO

(poly isoprene block rich in 1,4 microstructure)

1,4-rich microstructure:



Composition:

Mn x 10 ³ PIP-b-EO	Mw/Mn (PDI)
95.0-b-9.0	1.05

Synthesis Procedure:

Poly(Isoprene 1,4 addition or 1,2 addition)-b-ethylene oxide) can be prepared by the different routes as reported in the literature (Ref: *Macromolecules* 1996, 29, 6994). The direct synthesis of diblock copolymer using lithium counter ion in the presence of **Phosphazene Base t-BuP₄** is interesting as reported in *Macromolecules*, **32** (8), 2783 -2785, 1999. These polymers can also be successfully synthesized using different end functionalized polymers as investigated in our laboratory which are proprietary.

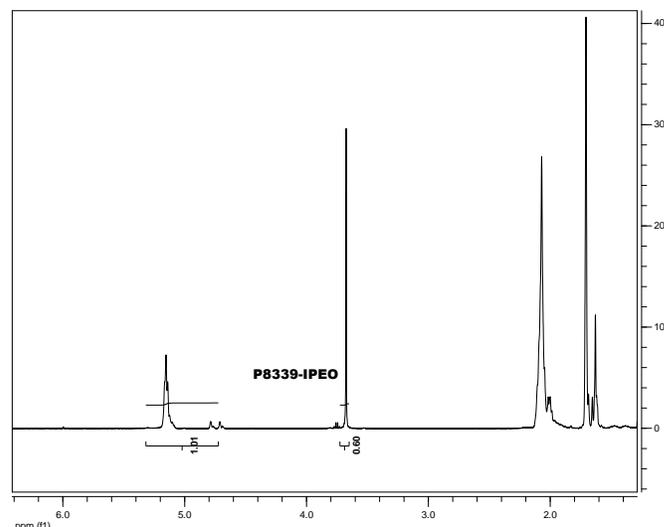
Characterization:

OH terminated isoprene was analyzed by size exclusion chromatography (SEC) to obtain the molecular weight and polydispersity index (PDI). The final block copolymer composition was calculated from ¹H NMR spectroscopy by comparing the peak area of the vinylic butadiene protons at about 5.4 ppm with the ethylene oxide protons at 3.6 ppm. Block copolymer PDI is determined by SEC.

Solubility:

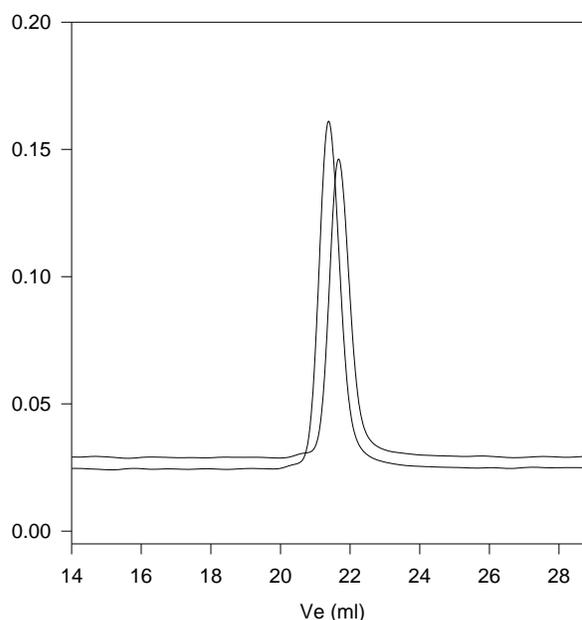
Poly(isoprene-b-ethylene oxide) is soluble in THF, CHCl₃, and toluene. The polymer has variable solubility in hexane, methanol, ethanol and water depending on its composition.

¹H NMR spectrum of the polymer:



SEC profile of the block copolymer:

P8339-IPEO Poly isoprene rich in 1,4 addition



Size exclusion chromatography of poly(Isoprene-b-ethylene oxide):

- polyisoprene (1,4 addition) M_n=95000, M_w=99000, PI=1.05
Solution Viscosity in THF at 35 oC: 0.858dl/g
Rg: 13.92 nm dn/dcin THF at 35 oC: 0.125 ml/g
- Block Copolymer PIP(95000)-b-PEO(9000), PI=1.05
Composition from H NMR

Thermal analysis of the sample# P8339-IPEO

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature (T_g).

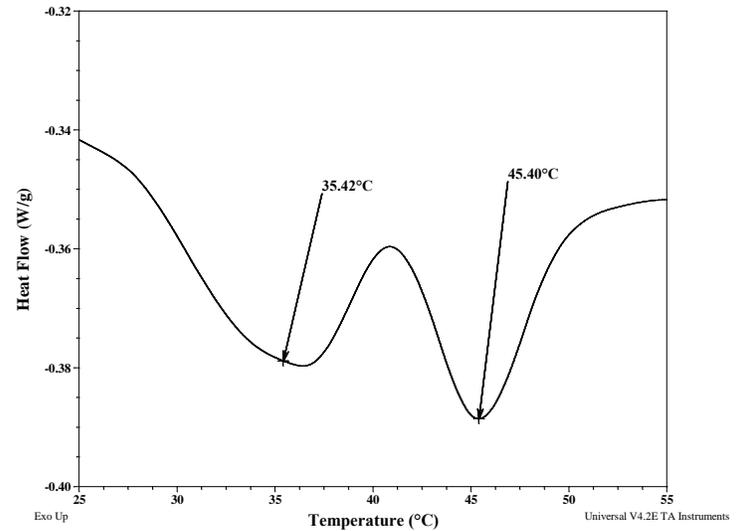
Melting and crystallization curve for the sample

The melting temperature (T_m) was taken as the maximum of the endothermic peak where as the crystallization temperature (T_c) was considered as the minimum of the exothermic peak.

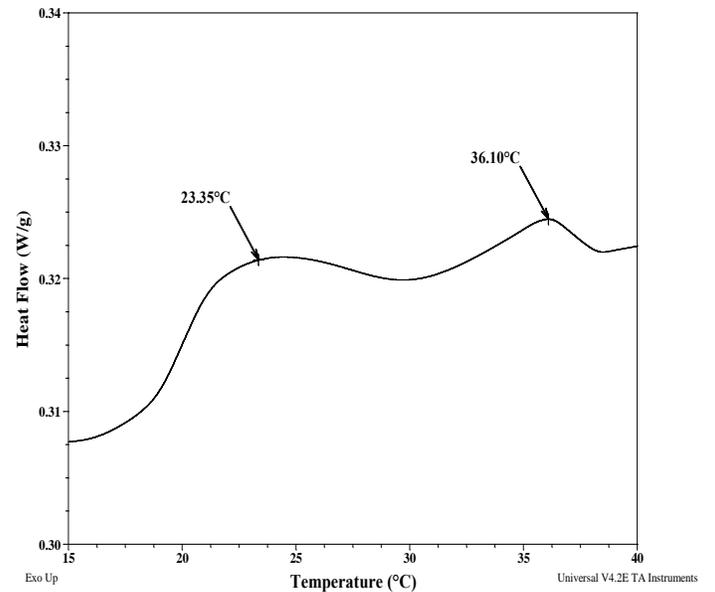
Thermal analysis results at a glance

Sample	T_m (°C)	T_c (°C)	T_g (°C)
EO	35 & 45	23 & 36	-
Ip	-	-	-62

Melting curve for PEO block:



Crystallization curve for PEO block:



Thermogram for the sample

