

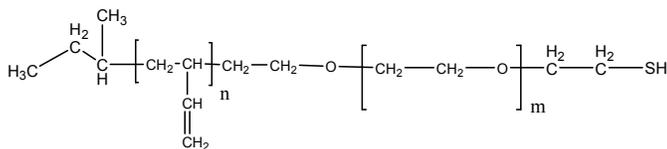
Sample Name: Thiol terminated Poly(butadiene-b-ethylene oxide)

*Poly butadiene rich in 1,2 or 1,4 microstructure*

Sample #: P10804-BdEOSH

*(poly butadiene block rich in 1,2 microstructure)*

Structure of 1,2-rich microstructure about 95%:



**Composition:**

Mn x 10 <sup>3</sup> Bd-b-EO	Mw/Mn (PDI)	% 1,2 addition Butadiene
2.5-b-1.3	1.09	95%

**Synthesis Procedure:**

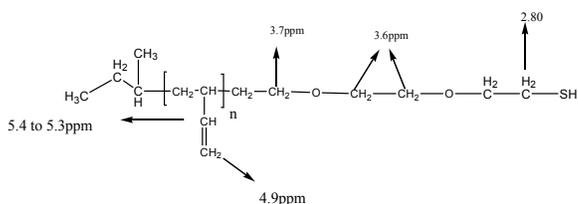
Poly(butadiene(1,4 addition or 1,2 addition)-b-ethylene oxide) can be prepared by the different routes as reported in the literature (ref: *Macromolecules* 1996, 29, 6994). The direct synthesis of diblock copolymer using lithium counter ion in the presence of **Phosphazene Base t-BuP<sub>4</sub>** is interesting as reported in *Macromolecules*, **32** (8), 2783 -2785, 1999. These polymers can also be successfully synthesized using the different end functionalized polymers as investigated in our lab. These methodologies are proprietary. Terminal OH modified to thiol.

**Characterization:**

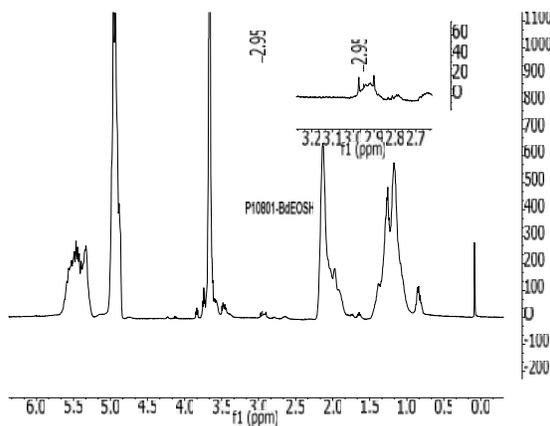
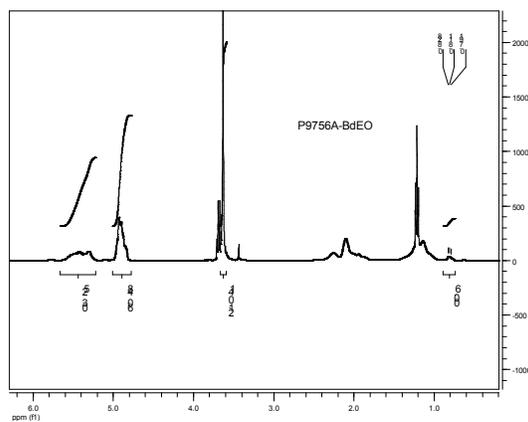
Polymer was analyzed by size exclusion chromatography (SEC) to obtain the molecular weight and polydispersity index (PDI). The final block copolymer composition was calculated from <sup>1</sup>H-NMR spectroscopy by comparing the peak area of the vinylic butadiene protons between about 5.0-5.4 ppm with the ethylene oxide protons at 3.6 ppm. Block copolymer PDI is determined by SEC. Note: The <sup>1</sup>H-NMR of 1,2-polybutadiene is composed of 1 proton signal at 5.4 ppm and 2 proton signals at 5.0 ppm. Signals due to vinylic 1,4-polybutadiene are also present at 5.4 ppm.

**Solubility:**

Polymer is soluble in THF, CHCl<sub>3</sub>, and toluene. The polymer has variable solubility in hexane, methanol, ethanol and water depending on its composition.

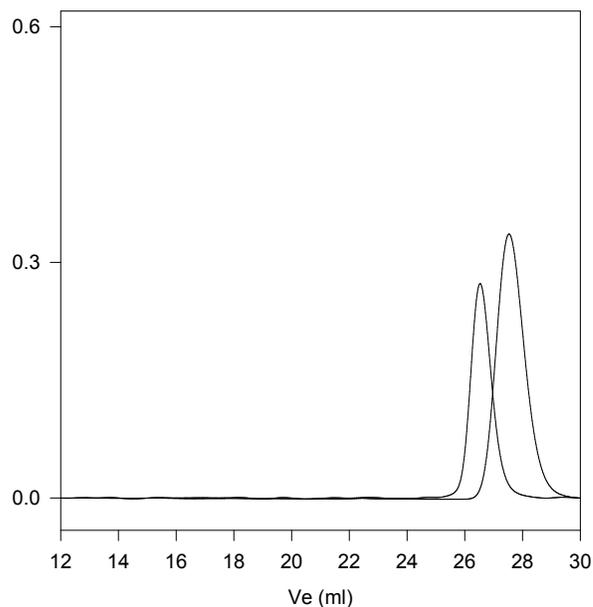


<sup>1</sup>H NMR spectrum of the sample (precursor)



**SEC Profile:**

**P10804-BdEOSH**



Size Exclusion Chromatogram of Poly(butadiene-b-ethylene oxide)

— Polybutadiene: M<sub>n</sub>=2500, M<sub>w</sub>=2700, M<sub>w</sub>/M<sub>n</sub>=1.07

— PBd-b-PEO: M<sub>n</sub> PBd(2500)-PEO(1300), M<sub>w</sub>/M<sub>n</sub>=1.09

The Mn of PEO is calculated from NMR results,