Sample Name: Polyisoprene epoxy

(1,4 rich addition and 1,2 and 3,4 addition epoxy)

Sample #: P10963-IPEpoxy

Structure:

$$\begin{array}{c} & \overset{\text{CH}_3}{\longleftarrow} & \overset{\text{CH}_2}{\longleftarrow} & \overset{\text{CH}_3}{\longleftarrow} & \overset{\text{CH}_2}{\longleftarrow} & \overset{\text{CH}_3}{\longleftarrow} & \overset{\text{$$

Composition:

M _n x 10 ³	PDI
2.6	1.2
Epoxydation: 8 mol %	

Synthesis Procedure:

Polyisoprene epoxy (1, 4 addition rich) is obtained by living anionic polymerization of isoprene in polar media and followed by epoxidation.

Control epoxidation: preferential epoxidation of the branch 1,2 addition fractions

Characterization:

% of expoxidation was analyzed by NMR.

Thermal analysis:

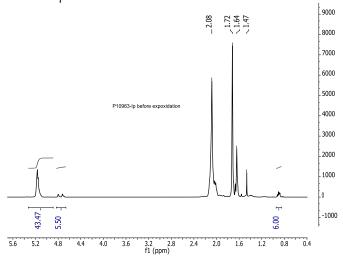
Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 20° C/min.The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature (T_g).

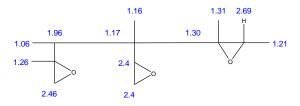
Solubility:

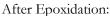
Polymer is soluble in THF, chloroform, toluene and in hexane.

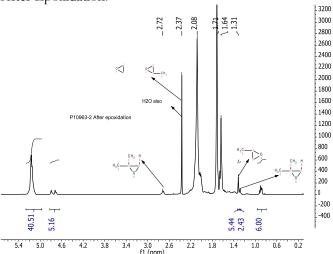
HNMR of homopolymer:

Before Epoxidation:



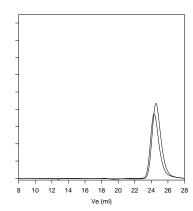






SEC of homopolymer:

P10963-Ip-Epoxy



Size exclusion chromatography of Polymer

- Polylsiprene before epoxidation: M_n=2,600 M_w: 3,100 PI=1.2
- ---- After Epoxidation: PI=1.2

DSC thermogram for the sample: