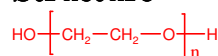


**Sample Name:** Poly(ethylene glycol)

or Poly ethylene oxide

**Sample #:** P5613-EG2OH or P5613-EO

**Structure:**

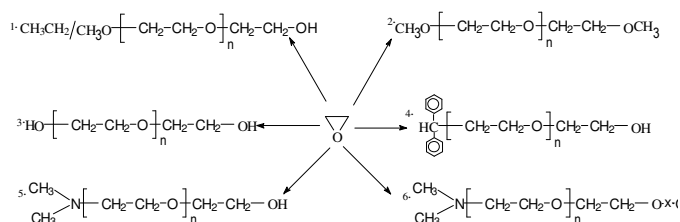


**Composition:**

$M_n \times 10^3$	PDI
450.0	1.4

**Synthesis Procedure:**

Poly (ethylene glycol) is obtained by living anionic polymerization and the reaction. Scheme of the polymerization is illustrated below:



Initiator System	Resulting Polymer
1) $\text{CH}_3\text{OCH}_2\text{CH}(\text{CH}_3)\text{OK}$	polyethylene glycol methyl ether
2) $\text{CH}_3\text{OCH}_2\text{CH}(\text{CH}_3)\text{OK}$	$\alpha$ , $\omega$ -term. methyl ether polyethylene glycol
3) $\text{KOCH}_2\text{CH}_2\text{OK}$	polyethylene glycol
4) $\text{CH}(\text{C}_6\text{H}_5)_2\text{CK}$	polyethylene glycol diphenyl ether
5) $(\text{CH}_3)_2\text{N}-\text{CH}_2\text{CH}_2\text{OK}$	methyl amino terminated PEG
6) $(\text{CH}_3)_2\text{N}-\text{CH}_2\text{CH}_2\text{OK}$	$\alpha$ -methyl amino $\omega$ -methyl ether term. PEG

**Characterization:**

By Size exclusion chromatography (SEC): Varian liquid chromatograph equipped with UV and refractive detector. SEC columns from Supelco were used with THF containing 2 vol%  $(\text{Et})_3\text{N}$  as the eluent. The molecular weights were determined using light scattering detector and viscosity detector. The molecular weights and the polydispersity indices were calculated.

An aqueous GPC column from Supelco(G5000 PWXL) was also used with 0.5 M acetic acid and 0.8 M  $\text{NaNO}_3$  as the eluent. It was kept at a constant temperature of  $50^\circ\text{C}$ . The flow rate was 1.0 ml/min. The column was calibrated with monodisperse poly(ethylene oxide) standards. The molecular weights and the polydispersity index of polyethylene oxide were calculated by using GPC software.

**Solubility:**

Poly(ethyl glycol) is soluble in toluene, THF, water and  $\text{CHCl}_3$ . The polymer is insoluble in hexane, ether, isopropanol and cold ethanol.

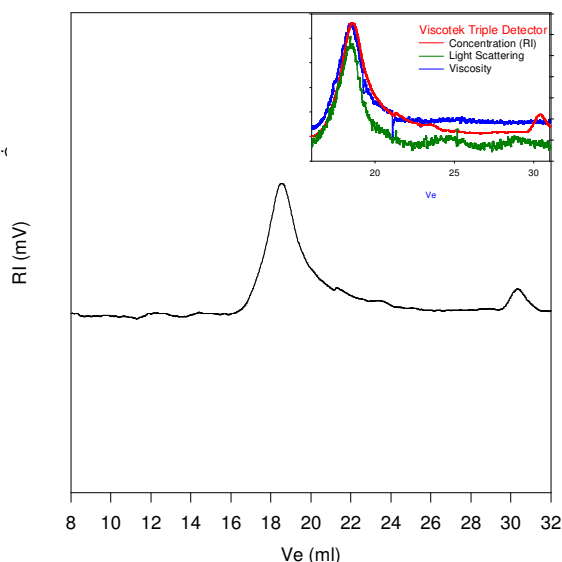
**Purification of the obtained polymer:**

Purification of the obtained polymer was carried out rigorously as follows to ensure the removal of the catalyst side product:

1. Dissolved the polymer in de-ionized distilled water to remove the any insoluble organic catalyst side product.
2. Polymer extracted from water with dichloromethane.
3. Polymer solution in dichloromethane was dried over anhydrous sodium sulfate.
4. Solution filtered and than passed through a column packed with basic  $\text{Al}_2\text{O}_3$ .
5. Solution concentrated on rota-evaporator
6. Solution precipitated in cold diethyl ether.
7. Dried under vacuum for 48h at  $38^\circ\text{C}$ .

**SEC of Sample**

**P5613-EO2OH or EG2OH**



Size Exclusion Chromatography of polymer:

—  $M_n = 450,000$ ,  $M_w = 630,000$ ,  $M_w/M_n = 1.4$   
 $dn/dc$  in THF: 0.067 ml/g  
 $R_{90}$ : 41.16nm

## Thermal analysis of the sample# P5613-Eg2OH

Thermal analysis of the samples was carried out on a TA Q100 differential scanning calorimeter at a heating rate of 10°C/min. The midpoint of the slope change of the heat flow plot of the second heating scan was considered as the glass transition temperature ( $T_g$ ).

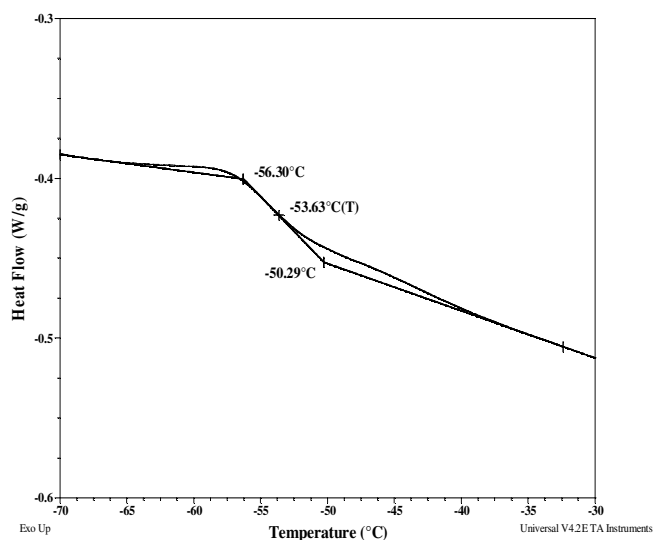
### Melting and crystallization curve for the sample

The melting temperature ( $T_m$ ) was taken as the maximum of the endothermic peak where as the crystallization temperature ( $T_c$ ) was considered as the minimum of the exothermic peak.

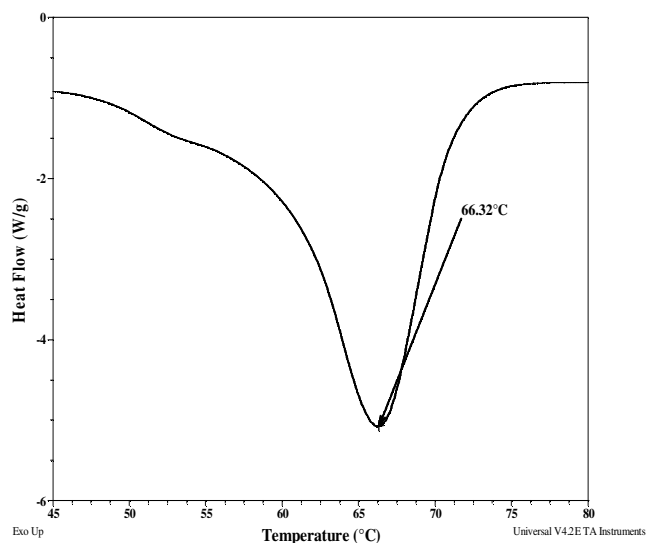
### Thermal analysis results at a glance

Sample	$T_m$ (°C)	$T_c$ (°C)	$T_g$ (°C)
EG2OH	66	47	-54

### Thermogram for the sample



### Melting curve for the sample:



### Crystallization curve for the sample:

